

Heterogenous Nucleation in Hard Spheres Systems

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Motivation Event Driven Molecular Dynamics

Time Driven MD simulation vs. Event Driven MD simulation

$$V(r) = \begin{cases} \infty & \text{if } r < \sigma \\ 0 & \text{if } r \geq \sigma \end{cases}$$

⇒ free flight between collisions

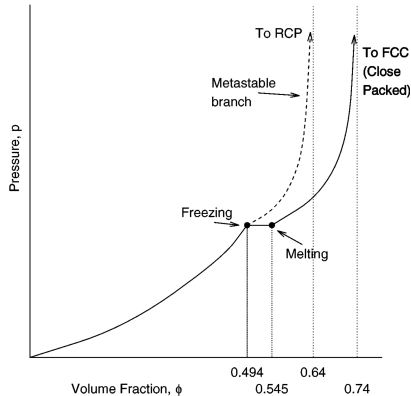
NVE ensemble.

$\tau = \frac{\sigma^2}{D}$ sets the timescale



B. J. Alder and T. E. Wainwright, J. Chem. Phys., 27, 1208 (1957).

Phasediagram for Hard Spheres



Md. Rintoul and S. Torquato, PRL 77, 20, 4198-4201 (1996).

Definition of the order parameter

Observables for the Local Bond Ordering:

In 3d:

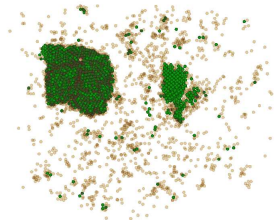
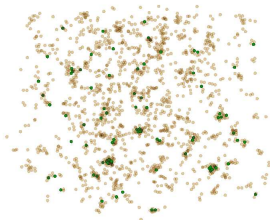
$$\bar{q}_{6m}(i) := \frac{1}{n(i)} \sum_{j=1}^{n(i)} Y_{6m}(\vec{r}_{ij}) \quad , r_{ij} < 1.4$$

where $Y_{6m}(\vec{r}_{ij})$ are the spherical harmonics (with $l=6$).

P. J. Steinhardt, D. R. Nelson, and M. Ronchetti. Phys. Rev. B, 28(2) (1983)

Detecting emerging crystallites

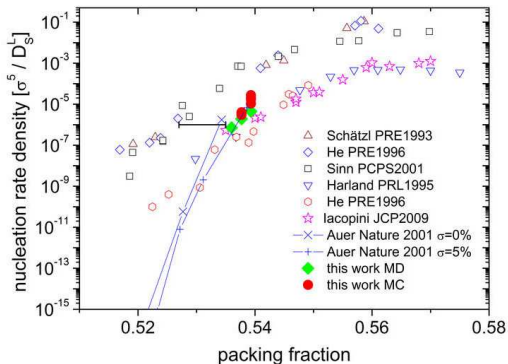
If particle i and n_b neighbouring particle j satisfy
 $\vec{q}_6(i) \cdot \vec{q}_6^*(j) > 0.7$



Crystalline particles (colorcoded "green"), $n_b > 10$.

Low symmetry cluster (LSC)(colorcoded "light brown"), $n_b > 5$.

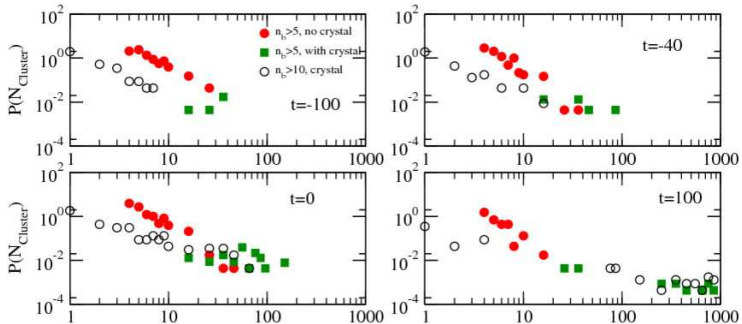
Nucleation rates



⇒ MD and MC simulations produce rates match the experimental results

T. Schilling, S. Dorosz et al. JPCM, 23, 19, 194120 (2011).

Precursor nucleation



⇒ Effective two step process. Precursor mediated nucleation

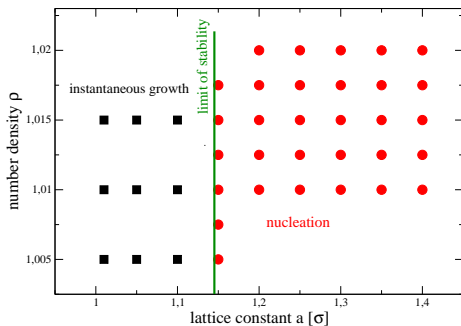
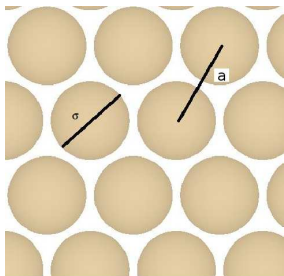
T. Schilling, S. Dorosz et al. JPCM, 23, 19, 194120 (2011).

Motivation

- Will the substrate induce different nucleation pathways?
- Where does the nucleation happen?
- What are the consequences of the mismatch between substrate and equilibrium crystal lattice constant?
- What is the crystal structure of the nucleus?
- How does the substrate change the nucleation rate?

Setup of the system

Super-saturated fluid of hard spheres in contact with a triangular substrate.



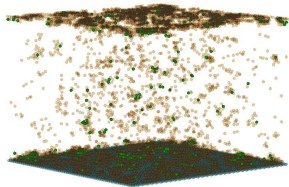
$N = 220200$ (216000 bulk + 4200 substrate) particles

$N/V = 1.005$ ($\eta = 0.526$) – 1.02 ($\eta = 0.534$)

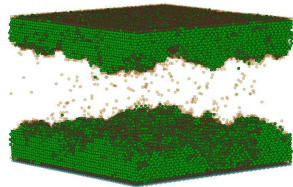
Corresponding chemical potentials $-\Delta\mu \simeq 0.50 - 0.54 k_B T$

Immediate wetting

$$a < a_{sp}$$



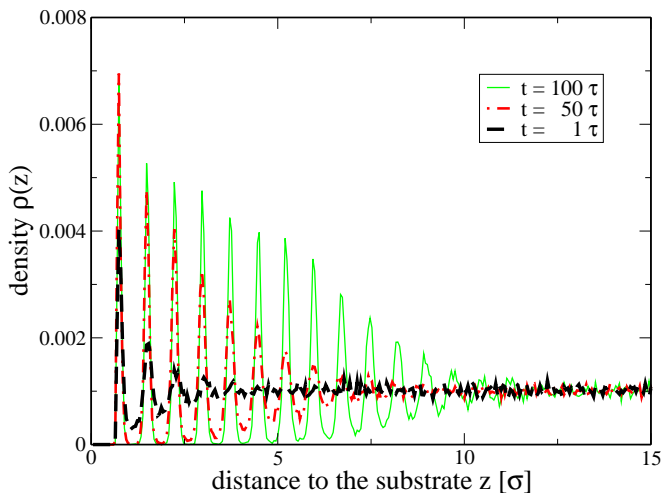
$$t = 6\tau$$



$$t = 70\tau$$

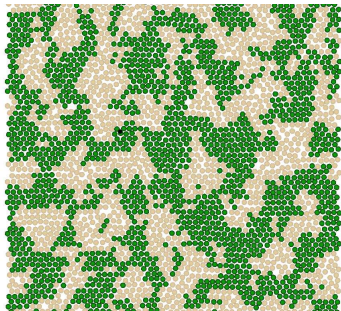
S.D. and T. Schilling, J. Chem. Phys. 136, 044702 (2012).

Vertical density profile

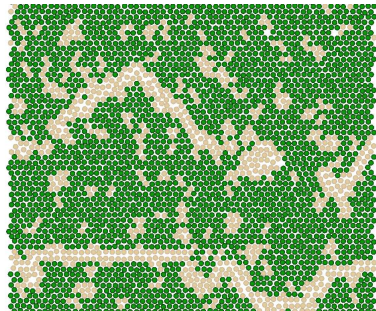


The first layer

at $t > 150\tau$



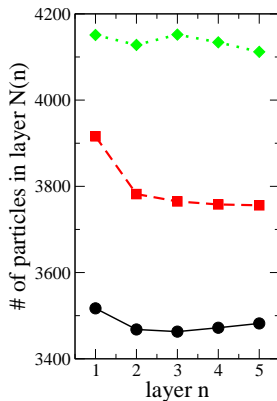
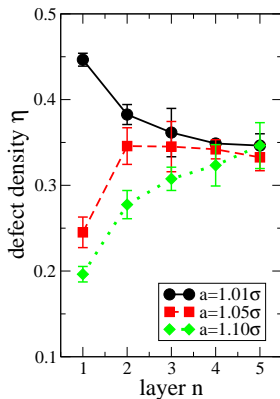
$$a = 1.01\sigma$$



$$a = 1.1\sigma$$

S.D. and T. Schilling, J. Chem. Phys. 136, 044702 (2012).

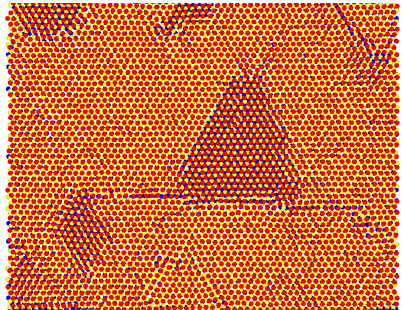
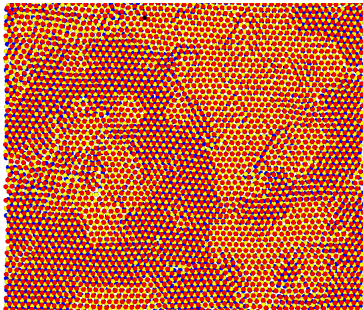
Defect density



⇒ Induced defects are compensated in the first 3 layers

S.D. and T. Schilling, J. Chem. Phys. 136, 044702 (2012).

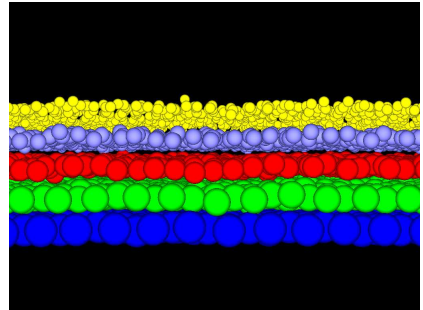
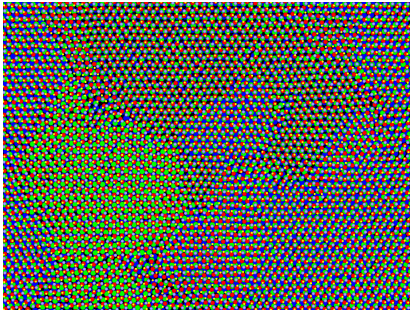
3 layer stacking



⇒ Domains of ABA and ABC structure

S.D. and T. Schilling, J. Chem. Phys. 136, 044702 (2012).

5 layer stacking



⇒ Crystal grows in random hexagonal closed packing (RHCP)

Distinguishing fcc/hcp and bcc

more detailed structure analysis with w_4 and w_6

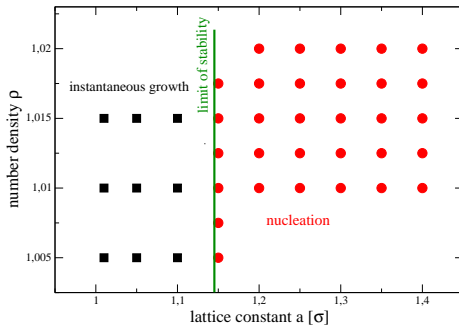
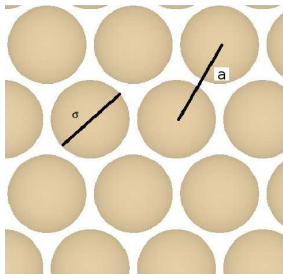
$$w_l = \frac{\sum_{m_1+m_2+m_3=l} \binom{l}{m_1 \ m_2 \ m_3} q_{lm_1} q_{lm_2} q_{lm_3}}{(\sum_{l=-m}^m |q_{lm}|^2)^{3/2}}$$

fcc	$-0.05 < w_6 < 0$	$w_4 < 0$
hcp	$-0.05 < w_6 < 0$	$w_4 > 0$
bcc	$0 < w_6 < 0.05$	$w_4 > 0$

\Rightarrow 85% of the crystal has fcc resp. hcp structure

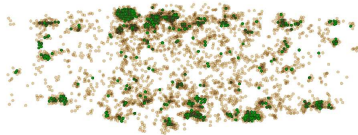
S. Jungblut and C. Dellago, 2011, J. Chem. Phys. 134, 104501.

Setup of the system



Nucleation at the wall

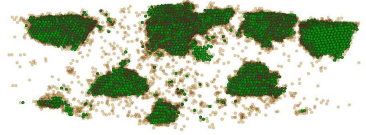
$$a > a_{sp}$$



$$t = 6\tau$$

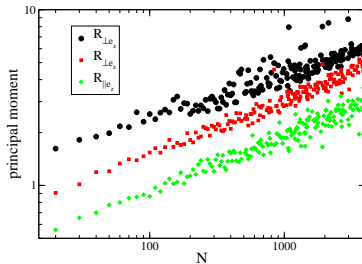
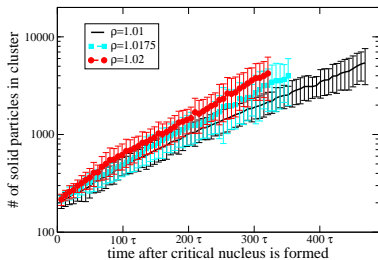
⇒ Droplet formation on the substrate

S.D. and T. Schilling, 2012, J. Chem. Phys. 136, 044702.



$$t = 80\tau$$

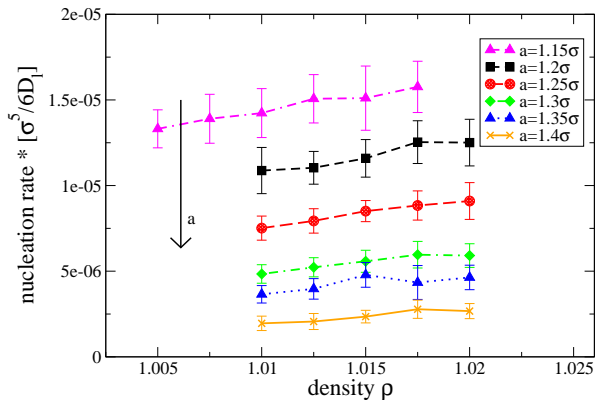
Droplet characterization



\Rightarrow non-spherical droplets even up to 4000 particles.

S.D. and T. Schilling, 2012, J. Chem. Phys. 136, 044702.

Nucleation rates



⇒ Decreasing nucleation rate with increasing mismatch to the substrate

S.D. and T. Schilling, 2012, J. Chem. Phys. 136, 044702.

Conclusion

- precursor mediated nucleation in homogeneous hard spheres systems
- two regimes of crystallization :
 - immediate full wetting of the surface
 - heterogeneous nucleation at the substrate

Acknowledgements



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